

Copper-containing Potassium-Alumina-Borate Glass

Structure and Nonlinear Optical Properties Correlation

Pavel Shirshnev¹, Nikolaj Nikonorov¹, Anastasija Babkina¹, Alexander Kim¹, Dmitriy Sobolev¹,
Ivan Kislyakov^{1,2}, Svjatoslav Povarov^{1,2}, Inna Belousova^{1,2} and Elena Kolobkova¹

¹Department of Optical Materials and Technology, ITMO University, Birzhevaya line 4, Saint-Petersburg, Russia

²Laboratory of Laser Physics, Federal State Unitary Scientific and Industrial Corporation "Vavilov State Optical Institute",
Birzhevaya line 12, Saint-Petersburg, Russia

Keywords: Nanocrystals, Nonlinear Optics, Glass Pottery.

Abstract: The paper describes the technology of obtaining potassium-alumina-borate glass with nanocrystals of copper chloride. Glass has transmission more than 75 % in visible range and nonlinear threshold less than 10^{-5} Joules (at the wavelength of 532 nm). Such low threshold can be explained by special structure of copper-containing nanophase in potassium-alumina-borate glass. In PAB glass low-melting temperatures of nanophase induce nonlinear effects in optics. In phosphate glass copper ions and clusters are responsible for optical nonlinearity.

1 INTRODUCTION

For the problems of photonics the special position among transparent in visible range materials is taken by crystal of copper chloride CuCl. CuCl is a wide gap semiconductor, in which direct transitions of electrons from the valence band to the conduction band is allowed (Lucas, Cowley, McNally, 2008). It has an extremely high exciton binding energy (190 meV), which allows to observe the exciton spectra at room temperature and also at higher temperatures (Efros, Onushchenko, Yekimov, 1985). The excitonic absorption intensity in near-UV (384 nm) is extremely high - more than 10^4 cm⁻¹ (Rivera, Murray, Hoss, 1967; Cordona, 1963; Cowley, 2011).

Copper chloride has a high nonlinear optical properties - due of two-photon absorption and of radiation-induced changing of refractive index (Yano, Goto, Itoh, 1996; Yasuaki, Makoto, Hideyuki, 1988; Kondo et al, 2000; Ichimiya et al, 2009).

It's known that temperature of melting of a CuCl is 426°C (Rivera, Murray, Hoss, 1967). CuCl has one big disadvantage - it's hygroscopic (Lucas, 2008), that's why it's not easy to use it in practical applications.

The solution of such problem is a formation of nanocrystals of copper chloride inside dielectric

matrix. The composite media based on dielectric matrices with dispersed semiconductor nanoparticles (glass pottery) have been extensively studied for several decades. Copper chloride nanocrystals was grown in silicate glass host by metastable phase separation (Dotsenko, Glebov, Tsekhomsky, 1998). Usually such glasses are photochromic (Dotsenko, Glebov, Tsekhomsky, 1998). Temperatures of melting point of copper chloride nanocrystals in this glass are reached 380°C and of crystallization 200°C (Onuschenko, Petrovskii, 1998; Valov, Leiman, 2009).

Potassium-alumina-borate glass host - is a perspective host with high liquation ability (Imaoka, Yamazaki, 1957; Kornilova, Petrovskii, Stepanov, 1980; Edelman et al, 2001). Copper chloride nanocrystals have been grown in potassium-alumina-borate glass host (Nikonorov, Tsekhomsky, Shirshnev, 2012) (PAB pottery). It was found, that such glass pottery is non-photochromic (Nikonorov, Tsekhomsky, Shirshnev, 2012). Also it was found, that such glass pottery has optical nonlinear properties (Kim et al, 2011). In (Golubkov et al, 2012) it was shown by small angle X-ray scattering method (SAXS) that PAB pottery has a complicated composition of nanophase which affects on melting processes of nanocrystals in glass. In (Babkina et al, 2014) it was shown, that copper clusters are formed

in PAB glass besides CuCl nanocrystals which affects in luminescence properties of glass pottery.

2 MATERIALS AND METHODS

In this paper three glass of different compositions were synthesized. Glass with the content (mol %) $17,5\text{K}_2\text{O} - 21\text{Al}_2\text{O}_3 - 46\text{B}_2\text{O}_3 - 8,9\text{NaCl} - 2,42\text{Cu}_2\text{O} - 0,23\text{SnO}_2 - 0,23\text{Sb}_2\text{O}_3 - 0,82\text{Na}_3\text{AlF}_6 - 2,9\text{P}_2\text{O}_5$ with a glass transition temperature of about 380°C was prepared as the basis for preparation of halidecontaining glass crystalline materials. The glass was synthesized in amounts of 100 g in a corundum crucible with the use of a quartz stirrer at a temperature of 1350°C for 2 h. To release mechanical stresses, the annealing was conducted in a stepwise regime, starting from 380°C (a fine annealing according to the program for this glass). In the following we denote it as glass 1.

In the second glass composition was (mol %): $20,12\text{K}_2\text{O} - 24,14\text{Al}_2\text{O}_3 - 52,89\text{B}_2\text{O}_3 - 1,39\text{Cu}_2\text{O} - 0,26\text{SnO}_2 - 0,26\text{Sb}_2\text{O}_3 - 0,94\text{Na}_3\text{AlF}_6$ and 1 g of $\text{C}_{12}\text{H}_{22}\text{O}_{11}$. Conditions of synthesis were the same. We denote it as glass 2.

Table 1: Properties of samples.

name of sample	Glass host	Nanophase type	details of annealing
1	PAB	CuCl	410°C 10 hours
2	PAB	copper-containing nanoclusters	Not
3	Phosphate	CuCl	Not
4	PAB	No	Not

The third one is (mol %) $20\text{BaPO}_3 - 5\text{Na}_3\text{AlF}_6 - 70\text{NaPO}_3 - 4\text{CuCl} - \text{HCl}$. We denote it as glass 3. It should be noted, that in this composition CuCl is in "clear" form.

The glass transition temperatures were determined on a differential scanning calorimeter STA 6000 (Perkin Elmer) and were 380°C for all samples.

The glass 1 sample were subjected to repeated heat treatment at 410°C for 10 h.

The absorption spectra of glass were measured by spectrophotometer Perkin Elmer Lambda 600. Luminescence was measured by LS-11 Perkin Elmer Spectrafluorimeter.

It should be noted, that measuring of absorption

spectra during heating - is a very effective method for processes studying of melting and crystallization in a copper chloride containing glass pottery (Rivera, 1967). Scheme of experimental setup is in figure 1. In this work avantes spectrometer Avaspec-1028 UV-VIS and balanced halogen-deuterium lamp was used. The measurement setup scheme is vertically designed.

The scheme of experimental setup of nonlinear optics characteristics studying is shown in figure 2. The laser beam is divided on two beams. The first is the reference beam, another one is directed to the sample and after to the photodetector. An energy of laser beam is varied by a system of filters from 10^{-9} J till 3×10^{-2} J. This range is achieved by using two cassette filters with a total transmittance range from 10^{-9} till 100%. The laser wavelength was $532,1$ nm, pulse length was 5×10^{-9} seconds. All measurements was made in focused beam.

Spectra and nonlinear properties were measured on samples with thickness 2 mm, thermal measurements were made on samples with thickness 0,1 mm.

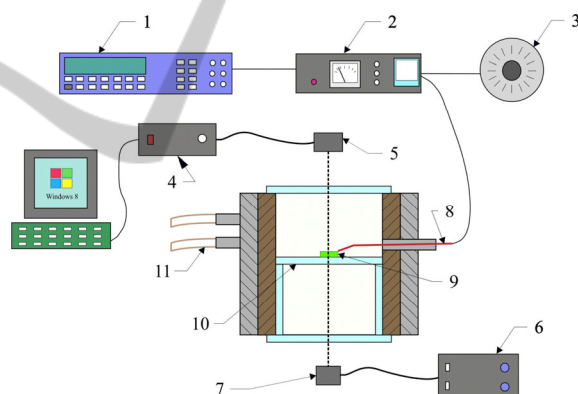


Figure 1: An experimental setup for absorption measurements: 1 - digital thermometer, 2,3 - heat controller, 4,5 - fiber spectrometer, 6,7 - deuterium lamp, 8 - thermal sensor, 9 - sample, 10 - quartz glass plate, 11 - cooling of thermocell.

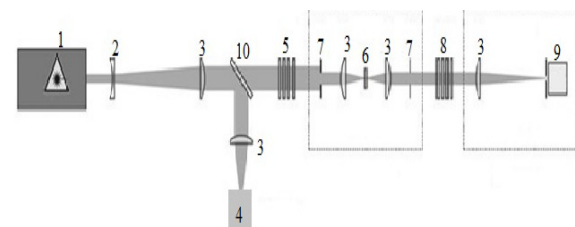


Figure 2: An experimental setup of nonlinear optical characteristics studying: 1 - laser, 2,3 - lenses, 4, 9 - photodetectors, 5,8 - blok of filters, 6 - sample, 7 - diaphragms, 10 - half-transmittable mirror.

3 RESULTS

3.1 Spectra of Optical Absorption

Spectra of absorption of glasses 1-3 are presented on the figure 3. Figure 3 shows absorption band are at 365-384 nm in glass 1 and 3. This band was identified as CuCl excitonic absorption (Dotsenko, 1998). There is no absorption at this wavelength range for the glass 2, even after heat treatment.

3.2 Spectra of Photoluminescence

Spectra of luminescence was measured for glasses 1-3 (figure 4). The excitation wavelength is 365 nm. From the figure 4 one can see that the luminescence of glasses 2-3 is identical. For the glass 1 the luminescence is not observed.

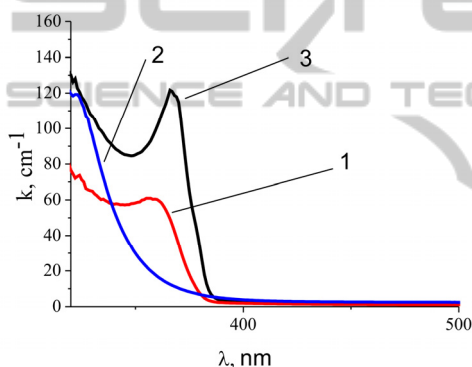


Figure 3: Spectra of optical absorption of glasses 1-3 (digits in figure the same as number of glasses).

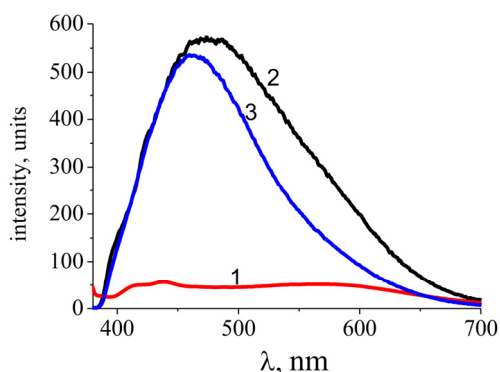


Figure 4: Spectra of photoluminescence for glass 1-3 (digits are the same as numbers of glasses).

3.3 Thermal Excitonic Spectroscopy

The dependence of excitonic absorption on temperature (glasses 1,3) is given in the figure 5 (the

thickness of samples 0,1 mm). As it can be seen, the temperature of excitone destruction for glasses 1 and 3 is really different (the difference is more than 200°C).

3.4 Nonlinear Properties Results

The dependence of optical transmission on energy of pulse for glasses 1-3 is shown in figure 6. It can be seen, that for the blank glass 1 there is no nonlinear effects. Nonlinear absorption characteristics of glasses 2 and 3 are the same, and the strongest effect is obtained on glass 1 (copper chloride containing PAB glass).

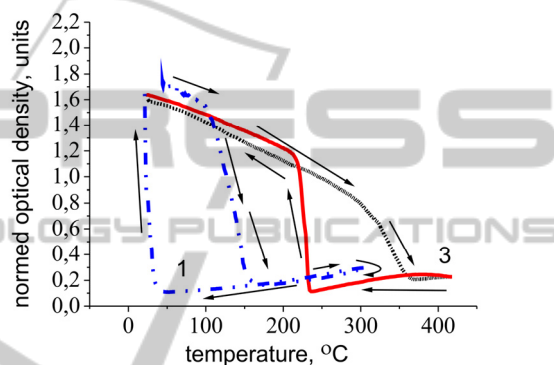


Figure 5: Excitonic absorption intensity of glass 1,3 at different temperatures.

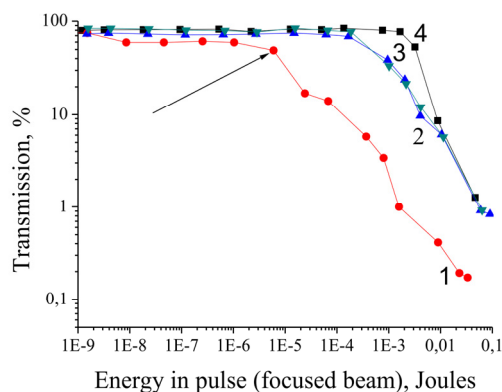


Figure 6: Nonlinear optical limiting on different samples (thickness of glass 2 mm): 1 – glass 1, 2 – glass 2, 3 – glass 3, 4 – blank (without CuCl) glass 1. Arrow shows the starting point of optical limiting process.

4 DISCUSSION

Let's discuss figure 5. As mentioned above, first glass had the lowest threshold of nonlinear limiting, first glass without nanocrystals had the highest one.

As we can see from absorption spectra of glass 3 (figure 3) nanocrystals of copper chloride were formed in glass. The relative concentration of CuCl in glass 1 and 3 can be estimated by absorption spectra intensity. So, for the glass 3 it's two times more than for the glass 1. So, we can say that the concentration of CuCl in sample 3 it's two times more than for the sample 1, but the strongest nonlinear optical limiting effect is in glass 1. Why? Let's discuss the figure 5. The temperature of full disappearing of excitone in glass 1 is less than 150°C, for glass 3 it's 420°C. It was shown in (Golubkov et al, 2012), that there are two characteristic temperatures of nanophase melting in PAB glass. One of that melting points is near 150°C another one is at 270°C. So, if we compare processes of nanophase melting in PAB glass with similar phenomena in silicate glasses with CuCl nanophase (Dotsenko, Glebov, Tsekhomsky, 1998), we can say, that temperature of excitone absorption disappearing at 150°C is the point of CuCl nanocrystals melting in glass 1. Point of CuCl melting in glass 3 is 361°C. It is known (Dotsenko, Glebov, Tsekhomsky, 1998; Golubkov et al, 2012) that CuCl containing nanophase can include another crystal components which can affects on value of melting point of nanophase. For CuCl-KCl eutectics such temperatures can be 150°C. So, in glass 3 CuCl nanophase can be "clear" and in glass 1 – nanophase can contain mixed eutectic crystals of CuCl and K₂CuCl₃.

The phase transition in such heterogeneous material as a PAB glass with CuCl nanocrystals can induce the optical limiting effect because of difference of refraction index of liquid CuCl-KCl solution and glass host. Also, the melting of nanophase can induce the growth of scattering and growth of optical absorption in visible range (including 532 nm).

In figure 5 the curves for glass 3 and 2 are identical. That could be mean, that the mechanisms of nonlinear limiting are the same in this glasses. As it shown in figure 4, the luminescence bands in glass 2 and 3 are the same. We suggested that glass 2 and 3 has Cu⁺ ions and (maybe) Cu_n⁰ clusters after comparing luminescence spectra with spectral luminescence data at (Babkina et al, 2014). In the article (Qiaohong et al, 2009) authors has shown that metal clusters in glass can induce nonlinear optical effects.

5 CONCLUSIONS

The new material with low-limit optical threshold was synthesized – a PAB glass with copper chloride nanocrystals. Numerical value of threshold was 5x10⁻⁶ Joules. This threshold is due of special structure of nanophase of such glass, which contains CuCl and K₂CuCl₃ nanocrystals. Such complicated structure of nanophase gives low points of melting of nanophase. That's why during laser irradiation of PAB-glass phase is melting and that's induces the appearance of additional color centers in glass. It was shown that copper clusters also induce nonlinear effects.

ACKNOWLEDGEMENTS

This work was financially supported by Russian Scientific Foundation (Agreement # 14-23-00136).

REFERENCES

- Lucas F., Cowley A., McNally P. J., 2008. Structural, optical and electrical properties of Co-evaporated CuCl/KCl films. *Physica Status Solidi -Vol.6. P.114*
- Efros, Al. L., Onushchenko, A. A. Yekimov, A. I., 1985. Quantum size effects in semiconductor microcrystals, *Sol. St. Comm., Vol.56, P.921*
- Rivera J., Murray L. A., Hoss P. A., 1967. Growth of coprus chloride single crystals for optical modulators, *Journal of crystal growth., Vol.1. - P.171-176.*
- Cordona M., 1963. Optical properties of the silver and cuprous halides, *Physical Review, Vol.129(1). P.69-78.*
- Cowley A. J., 2011. *Novel ultra-violet/blue optoelectronic materials and devices based on copper halides (CuHa)*, PhD thesys, Dublin city university school of electronic engineering
- Yano S., Goto T., and Itoh T. 1996. Excitonic optical nonlinearity of CuCl microcrystals in a NaCl matrix, *J. Appl. Phys. -Vol.79. P.8216.*
- Yasuaki M., Makoto Y., Hideyuki S. 1988. Optical nonlinearities of excitons in CuCl microcrystals, *Applied Physics Letters. - Vol.53. P.1527.*
- Kondo Y., Kuroiwa Y., Sugimoto N., Manabe T., Ito S. 2000. Ultraviolet irradiation effect on the third-order optical nonlinearity of CuCl-microcrystallite-doped glass, *J. Opt. Soc. Am. B -Vol.17.-P.548-554.*
- Ichimiya M., Ashida M., Yasuda H., Ishihara H., Itoh T. 2009. Room Temperature Degenerate Four-Wave Mixing Due to Ultrafast Radiative Decay of Confined Excitons *OSA/CLEO/IQEC P.138.*
- A. V. Dotsenko, L. B. Glebov, and V. A. Tsekhomsky, 1998. *Physics and Chemistry of Photochromic Glasses*, CRC Press LLC.

- A. A. Onuschenko, G. T. Petrovskii, 1998. Size effects embedded in phase transitions of semiconductor nanoparticles embedded in glass, *Journal of non-crystalline solids, embedded in glass. Vol.196. -P.73-78*
- P. M. Valov and V. I. Leiman, 2009. Size distribution of CuCl nanoparticles in glass in various stages of nucleation, *Phys. Solid State V.52, P.1703*.
- Imaoka M., Yamazaki T. 1957. Glass formation range of borate systems between a group elements. *Rep. Inst. Ind. Sci. Univ. Tokyo. vol. 6, p. 127-183*.
- Kornilova E. E., Petrovskii G. T. and Stepanov S.A. 1980. Unusual properties of iron ion clusters in poliboric glasses. *Dokl.Akad.Nauk SSSR, vol. 251 p. 409-413*
- I. S. Edelman, S. A. Stepanov, R. D.Ivantsov, T. V.Zarubina, E. E. Kornilova, A. D. Vasil'ev 2001. Borate Glasses with paramagnetic dopants – a new magneto optic material for the IR spectral range, *Glass Physics and Chemistry, Vol. 27, pp. 454-459*
- N. V.Nikonorov, V. A. Tsekhomskiy, P. S. Shirshnev 2012. Optical glass crystalline material with sharp boundary absorption in the UV region of the spectrum and its production method, *Patent of the Russian Federation, № 2466107*.
- A. A. Kim, N. V. Nikonorov, A. I. Sidorov, V. A. Tsekhomskii, P. S. Shirshnev 2011. Nonlinear optical effects in glasses containing copper chloride nanocrystals, *Technical Physics Letters, V.37, P.401-403*
- V. Golubkov, A. Kim, N. Nikonorov, V. Tsekhomskii, P. Shirshnev 2012. Precipitation of nanosized crystals CuBr and CuCl in potassium aluminoborate glasses, *Glass Physics and Chemistry, V. 38, P. 259-268*
- Babkina A. N., Nikonorov N. V., Shakhverdov T. A., Shirshnev P. S., Sidorov A. I. 2014. Luminescent thermochromism in potassium-alumina-borate glass with copper-containing molecular clusters at elevated temperatures, *Optical Materials. Vol. 116 pp. 773-777*.
- Babkina A. N., Nikonorov N. V., Sidorov A. I., Shirshnev P. S., Shakhverdov T. A. 2014. The effect of temperature on the luminescence spectra of potassium-aluminum borate and silicate glasses with copper(I) and silver ions, *Optics and spectroscopy. Vol. 116. pp. 84-90*.
- Li Qiaohong , Wu Kechen ,Wei Yongqin, Sa Rongjian, Cui Yiping, Lu Canguai, Zhu Jing and He Jianguang 2009. Second-order nonlinear optical properties of transition metal clusters $[MoS_4Cu_4X_2Py_2]$ (M = Mo, W; X = Br, I) *Phys. Chem. Chem. Phys., V.11, 4490-4497*